

Basic Electronics



Electronics

Introduction to Electronics

Lecture Contents



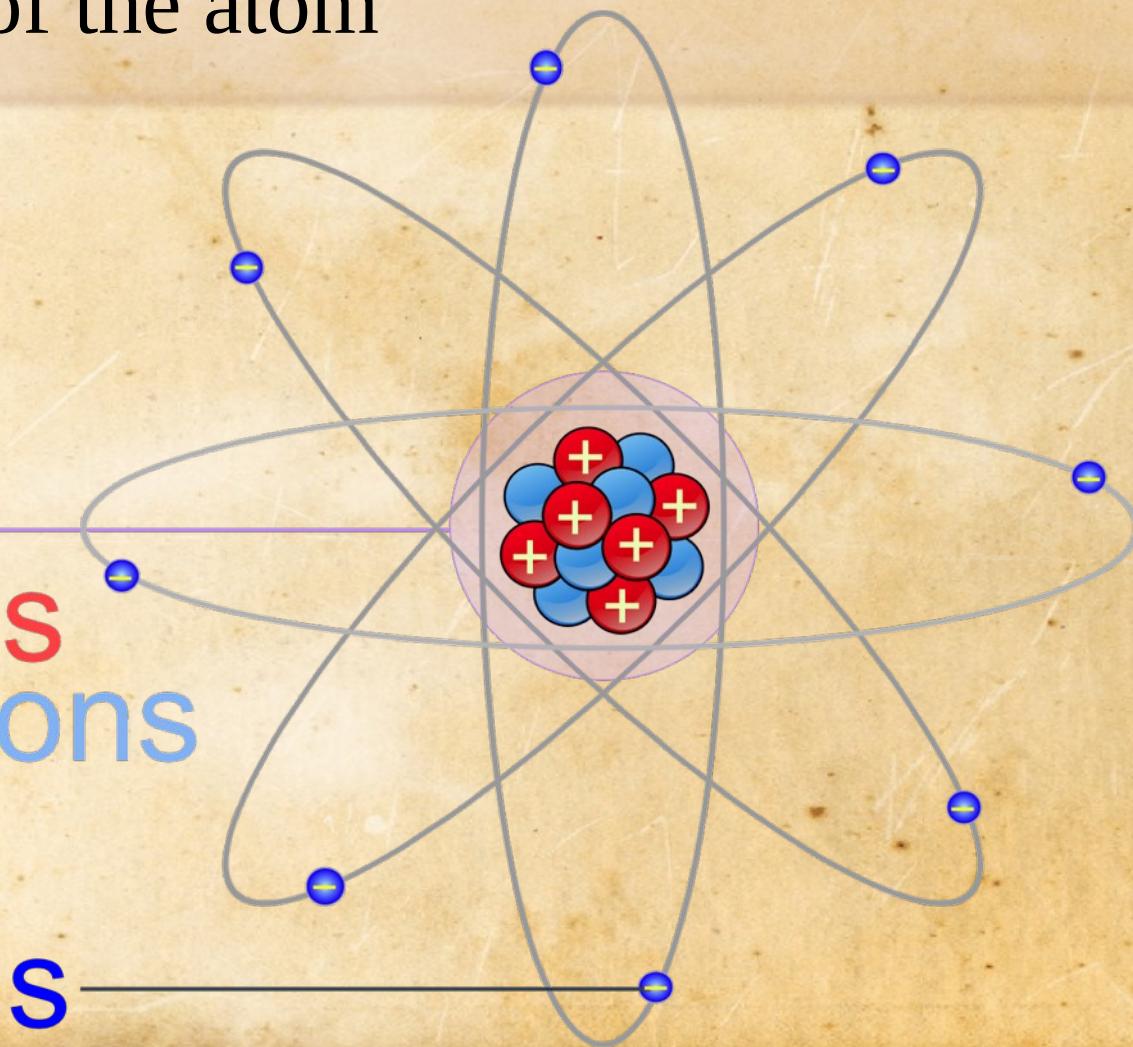
- Chemistry
 - Atoms, charges, electrons
 - Electron flow and current

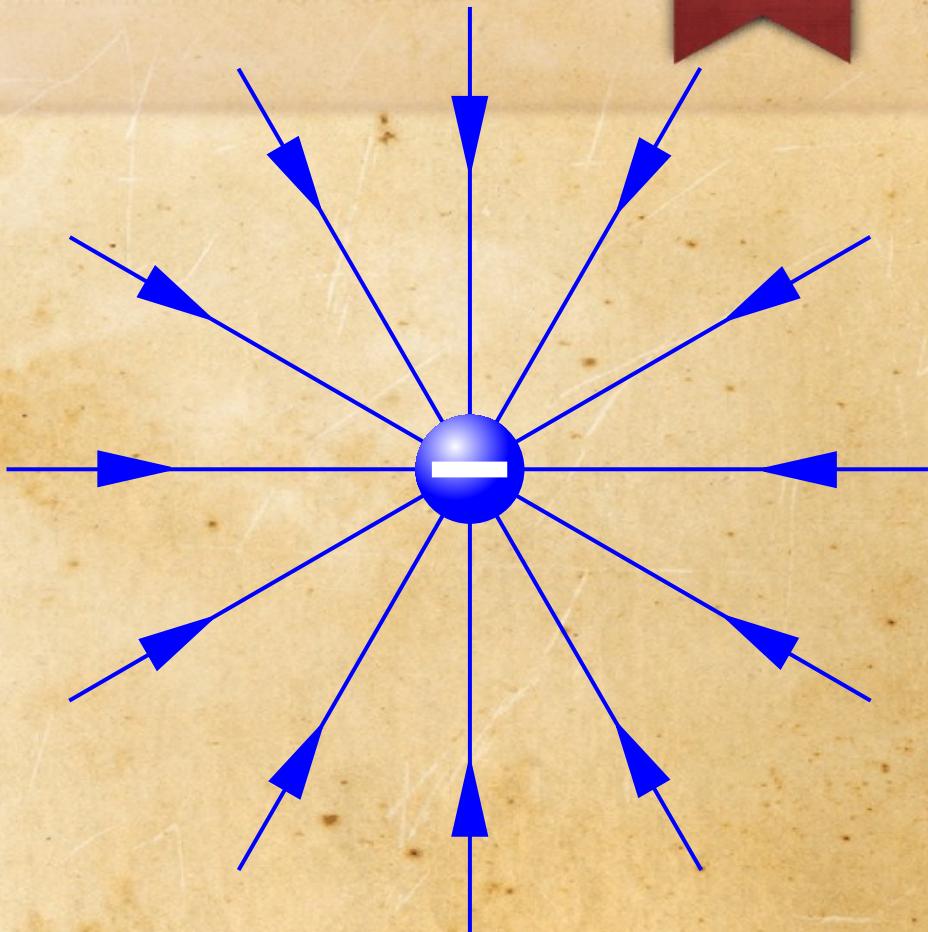
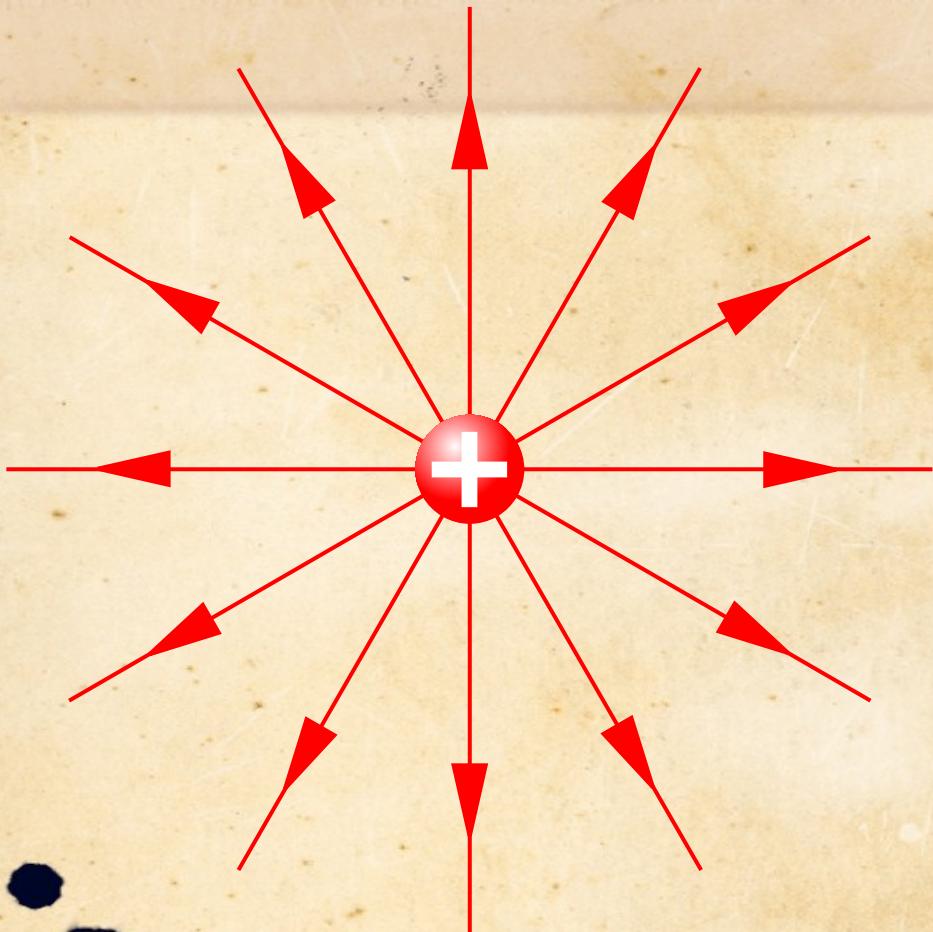
Visualization of the atom

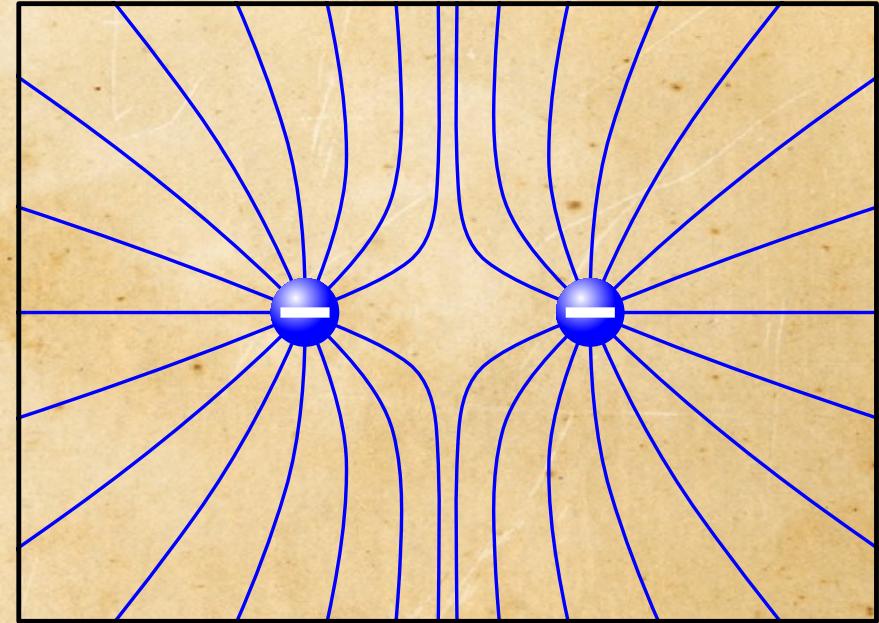
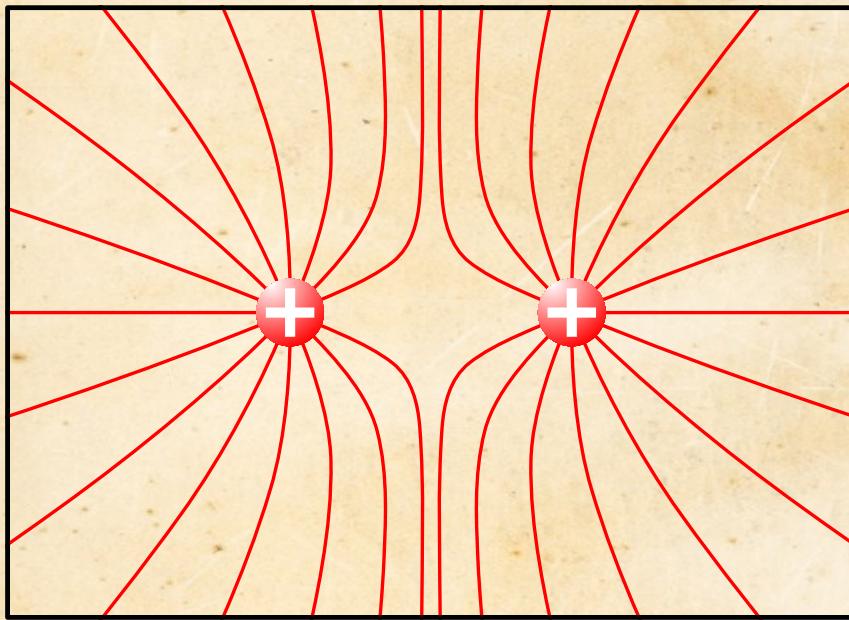


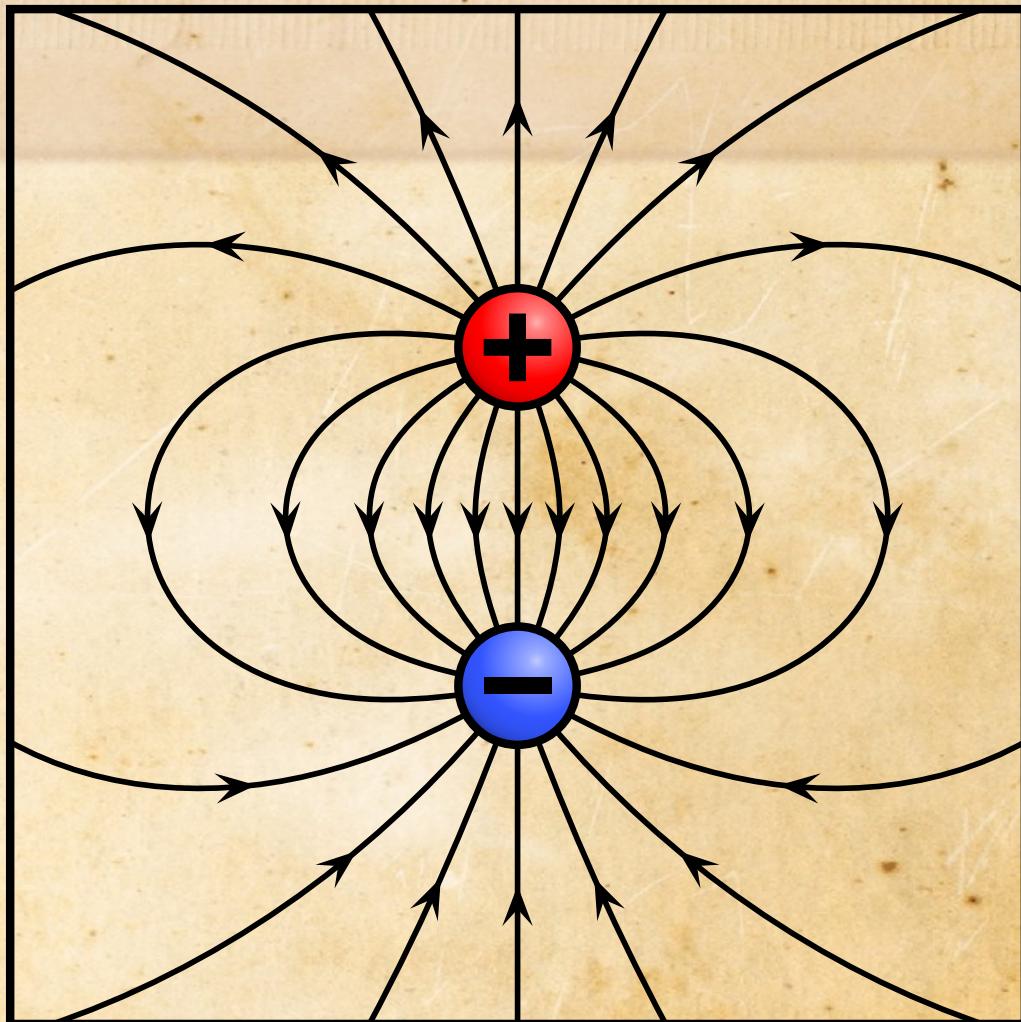
nucleus
protons
neutrons

electrons







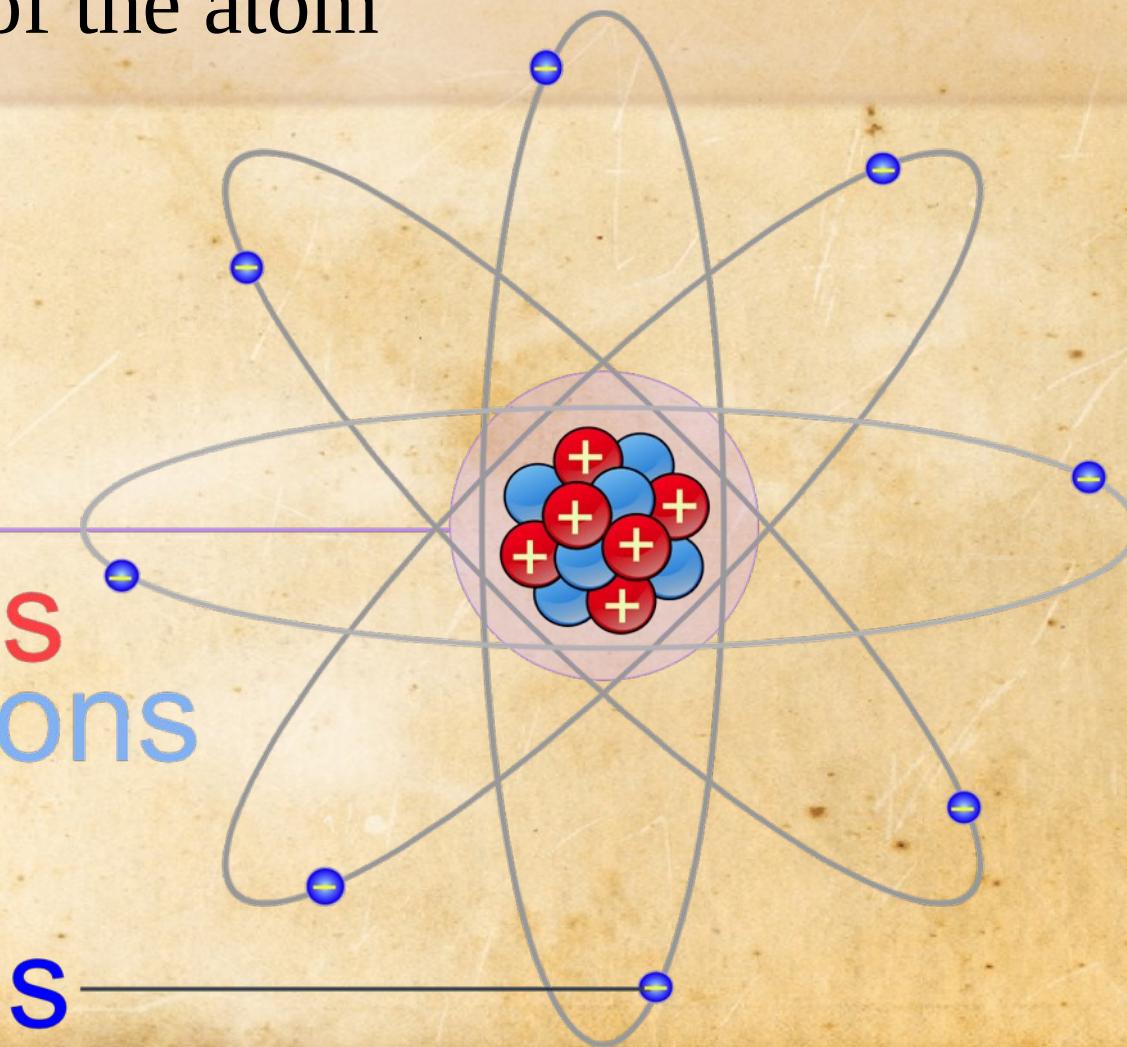


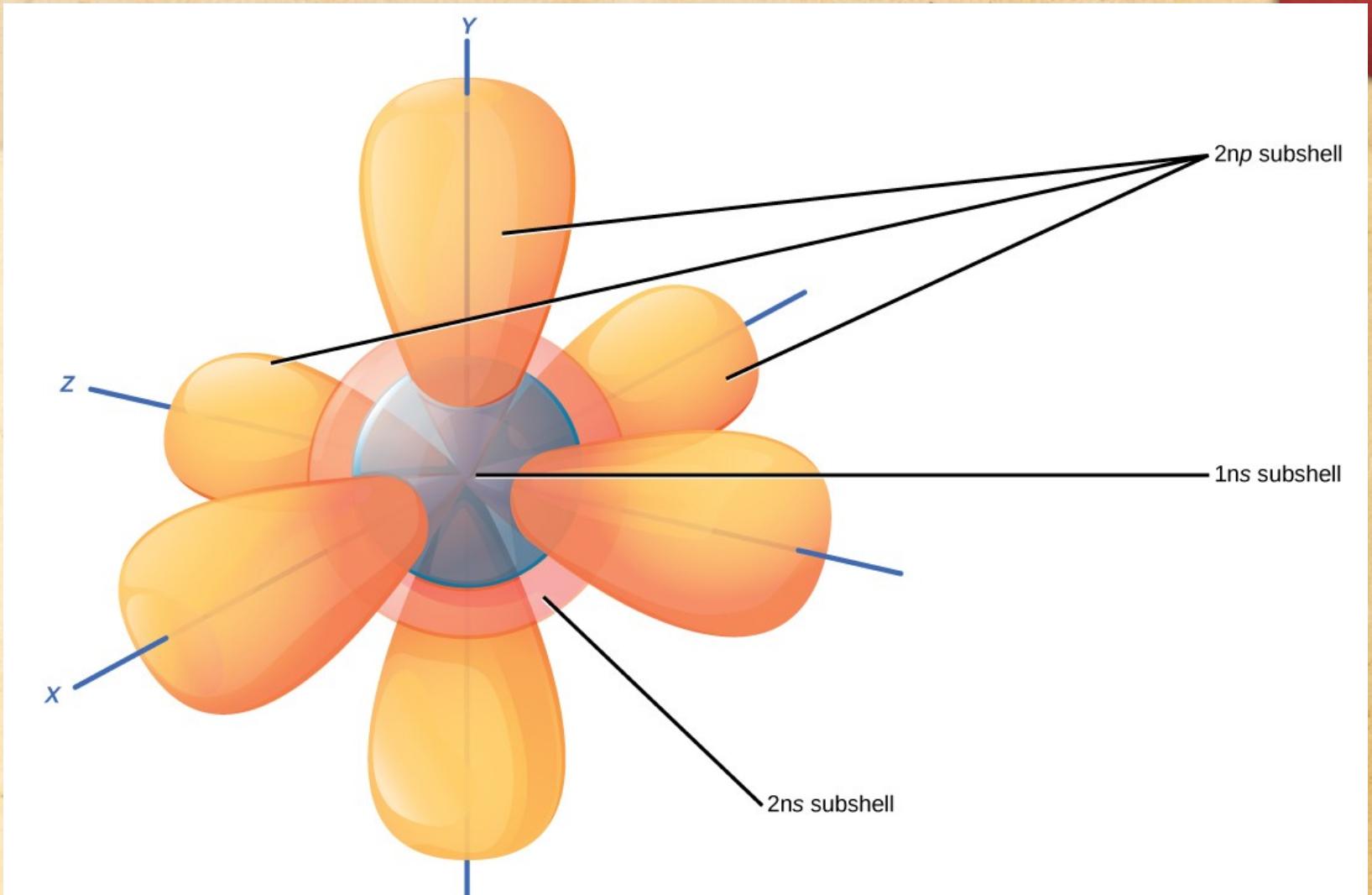
Visualization of the atom



nucleus
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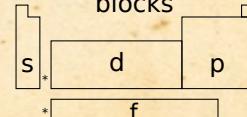




Periodic Table of the Elements

Group		Periodic Table of the Elements																						
Period	Element	Atomic Number	Symbol	Name	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18		
1	H	1	H	Hydrogen																	He	2		
2	Li	3	Be	Lithium	Beryllium																Helium	10		
3	Na	11	Mg	Sodium	Magnesium																Neon	18		
4	K	19	Ca	Potassium	Calcium																Ne	9		
5	Rb	37	Sr	Rubidium	Strontium																Ar	36		
6	Cs	55	Ba	Caesium	Barium																Kr	54		
7	Fr	87	Ra	Francium	Radium	*															Xe	86		
						*	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br			
						*	Scandium	Titanium	Vanadium	Chromium	Manganese	Iron	Cobalt	Nickel	Copper	Zinc	Gallium	Germanium	Arsenic	Selenium	Bromine	Krypton		
						*	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe		
						*	Yttrium	Zirconium	Niobium	Molybdenum	Technetium	Ruthenium	Rhodium	Palladium	Silver	Cadmium	Indium	Tin	Antimony	Tellurium	Iodine			
						*	Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn		
						*	Lutetium	Hafnium	Tantalum	Tungsten	Rhenium	Osmium	Iridium	Platinum	Gold	Mercury	Thallium	Lead	Bismuth	Polonium	Astatine			
						*	103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118		
						*	Lawrencium	Rutherfordium	Dubnium	Seaborgium	Bohrium	Hassium	Meitnerium	Darmstadtium	Roentgenium	Copernicium	Nihonium	Flerovium	Moscovium	Livermorium	Tennessine	Oganesson		

Electron configuration blocks



57	58	59	60	61	62	63	64	65	66	67	68	69	70
La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb
Lanthanum	Cerium	Praseodymium	Neodymium	Promethium	Samarium	Europium	Gadolinium	Terbium	Dysprosium	Holmium	Erbium	Thulium	Ytterbium
89	90	91	92	93	94	95	96	97	98	99	100	101	102
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No
Actinium	Thorium	Protactinium	Uranium	Neptunium	Plutonium	Americium	Curiום	Berkelium	Californium	Einsteinium	Fermium	Mendelevium	Nobelium



chemical symbol	C	6	atomic number (Z)
name	Carbon		

Transition metals in periodic table

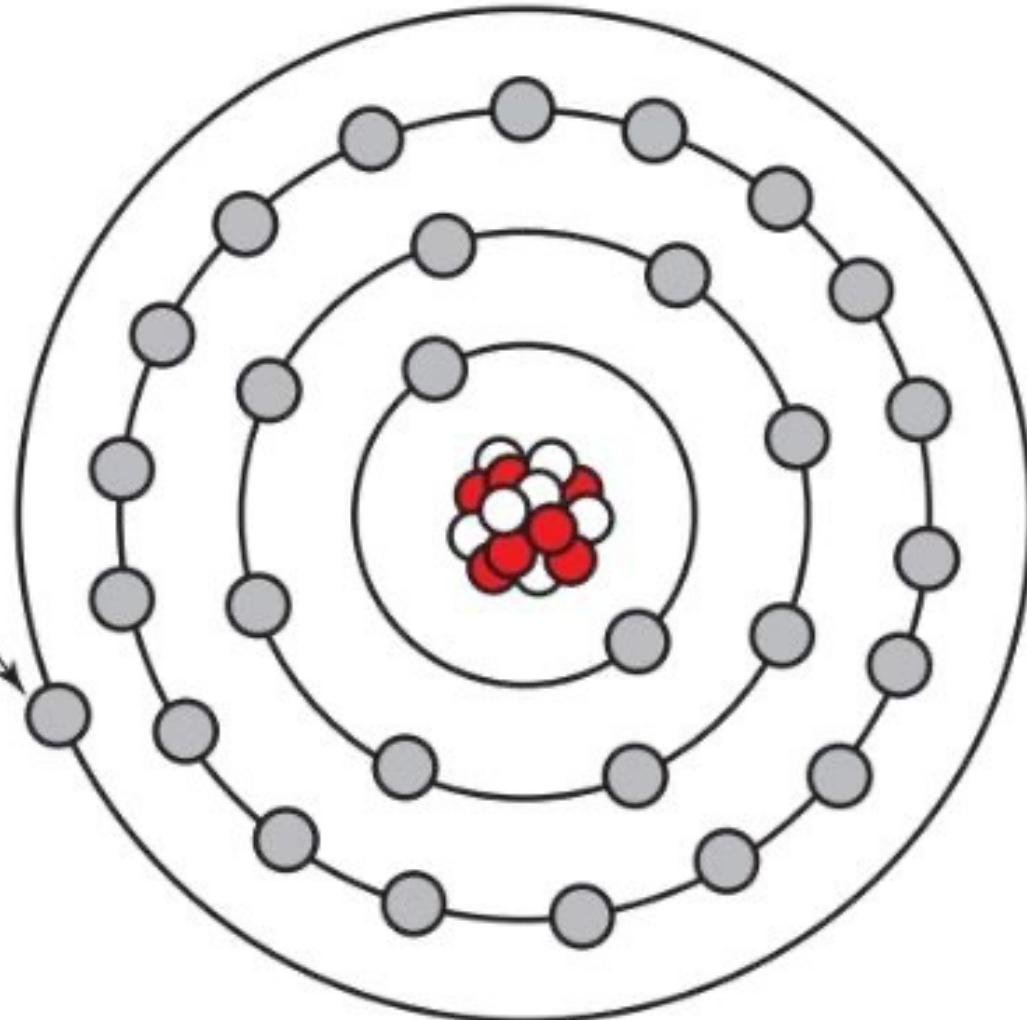
Atomic number → **1**
H ← Symbol
 Name → Hydrogen
 1.008 ← Atomic weight
 Electrons per shell → 1

Subcategory metals, nonmetals, and metalloids

Alkali metals	Lanthanides	Metalloids
Alkaline earth metals	Actinides	Reactive non metals
Transition metals	Post transition metals	Noble gases
		Unknown properties

1 H Hydrogen 1.008 1	2 He Helium 4.0026 2
3 Li Lithium 6.94 2-1	4 Be Beryllium 9.013 2-1
11 Na Sodium 22.98 2-8-1	12 Mg Magnesium 24.32 2-8-2
19 K Potassium 39.10 2-8-8-1	20 Ca Calcium 40.08 2-8-8-2
37 Rb Rubidium 85.47 2-8-18-8-1	38 Sr Strontium 87.62 2-8-18-8-2
55 Cs Cesium 132.91 2-8-18-8-1	56 Ba Barium 137.33 2-8-18-8-2
87 Fr Francium [223] 2-8-18-32-18-8-1	88 Ra Radium [226] 2-8-18-32-18-8-2
57 La Lanthanum 138.91 2-8-18-8-9-2	58 Ce Cerium 140.12 2-8-18-19-9-2
89 Ac Actinium [227] 2-8-18-32-18-9-2	59 Pr Praseodymium 140.91 2-8-18-21-8-2
90 Th Thorium 232.04 2-8-18-32-18-10-2	60 Nd Neodymium 144.24 2-8-18-22-8-2
91 Pa Protactinium 231.04 2-8-18-32-21-9-2	61 Pm Promethium [145] 2-8-18-23-8-2
92 U Uranium 238.03 2-8-18-32-22-9-2	62 Sm Samarium 150.36 2-8-18-24-8-2
93 Np Neptunium [237] 2-8-18-32-22-9-2	63 Eu Europium 151.96 2-8-18-25-8-2
94 Pu Plutonium [244] 2-8-18-32-25-8-2	64 Gd Gadolinium 157.25 2-8-18-25-9-2
95 Am Americium [243] 2-8-18-32-25-8-2	65 Tb Terbium 158.93 2-8-18-27-8-2
96 Cm Curium [247] 2-8-18-32-25-9-2	66 Dy Dysprosium 162.50 2-8-18-28-8-2
97 Bk Berkelium [247] 2-8-18-32-27-8-2	67 Ho Holmium 164.93 2-8-18-29-8-2
98 Cf Californium [251] 2-8-18-32-28-8-2	68 Er Erbium 167.26 2-8-18-30-8-2
99 Es Einsteinium [252] 2-8-18-32-29-8-2	69 Tm Thulium 168.93 2-8-18-31-8-2
100 Fm Fermium [257] 2-8-18-32-30-8-2	70 Yb Ytterbium 173.05 2-8-18-32-8-2
101 Md Mendelevium [258] 2-8-18-32-31-8-2	71 Lu Lutetium 174.97 2-8-18-32-9-2
102 No Nobelium [259] 2-8-18-32-32-8-2	103 Lw Lawrencium [266] 2-8-18-32-32-8-3

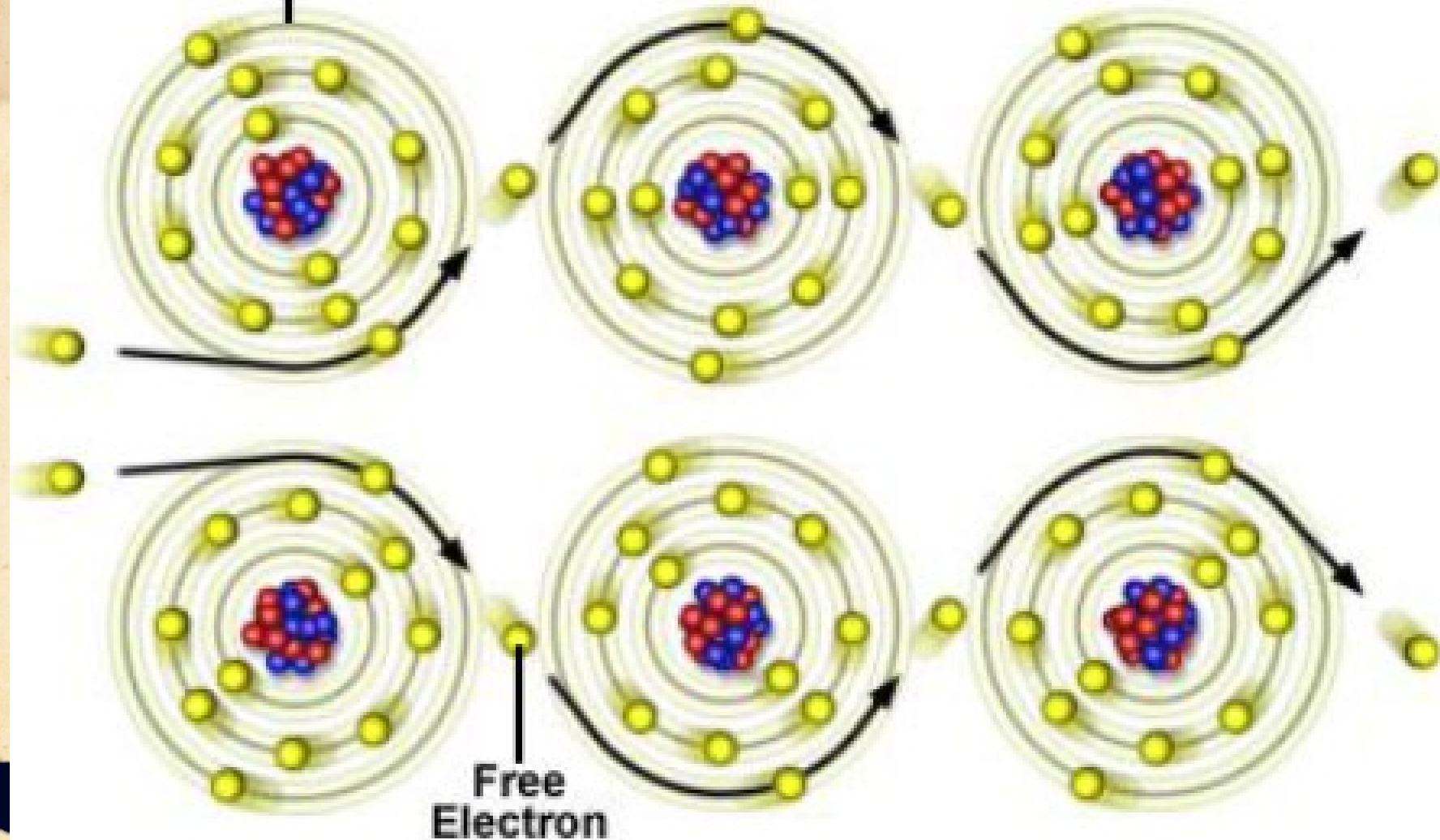
Single electron in
the outer orbit
of copper atom



Copper atom

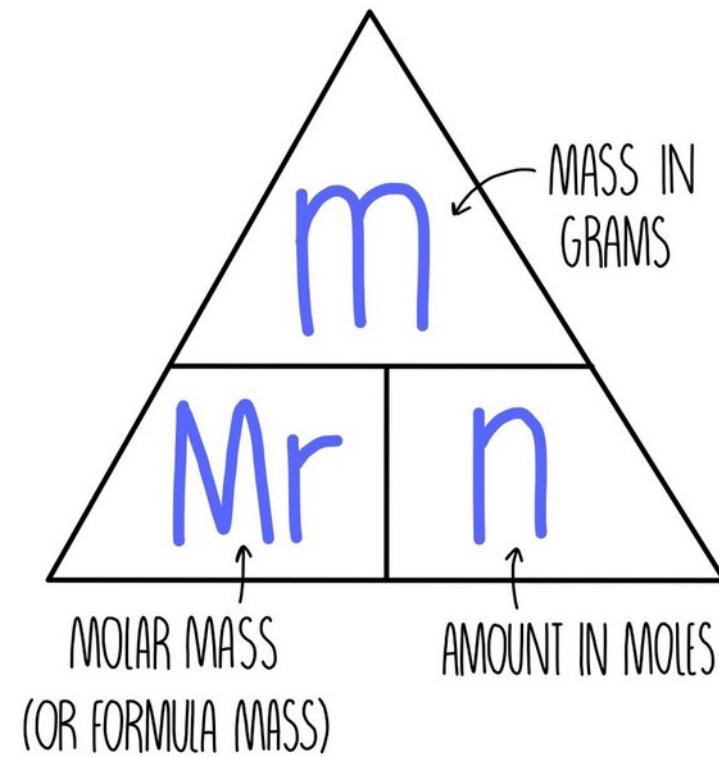
Valence Orbit

Direction of Current →



Chemistry – Avogadro's number

- N_A – Avogadro's number
 - $6.02214076 \times 10^{23}$ particles / mole

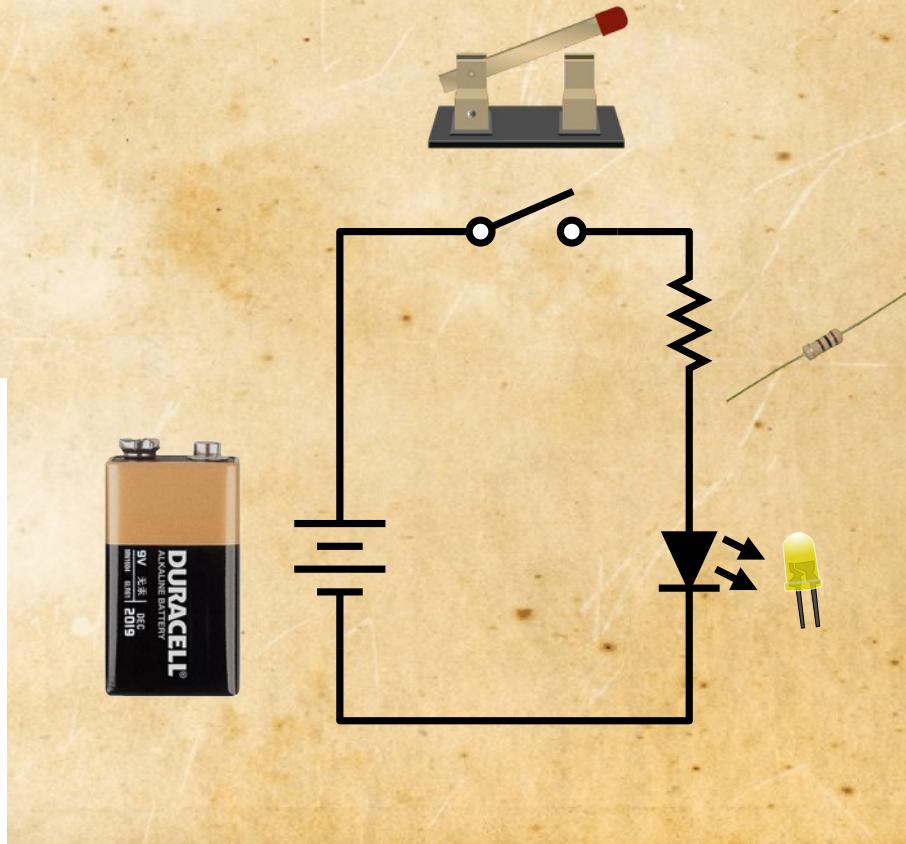
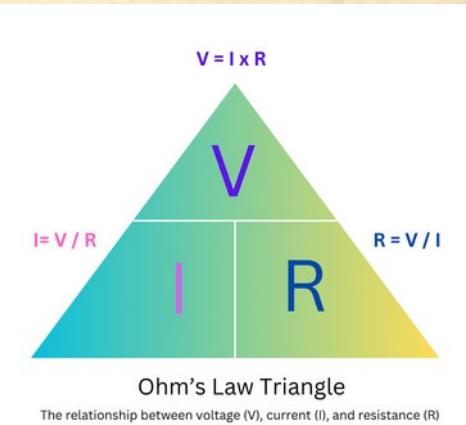


Electronics – Coulomb

- Avogadro's Number, $N_A = 6.02214076 \times 10^{23}$ particles / mole
- **Charge:** 1 Coulomb of charge per 6.24150975^{18} electrons
 - 1 electron has a charge of **$1.60217646 \times 10^{-19}$ Coulombs**
- **Current:** 1 Ampere = 1 Coulomb per second

Current and Voltage

- **Current** is the amount of charge (the number of electrons) flowing through the circuit per unit time
- **Voltage** is the strength with which those electrons are pushed.



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